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States of matter | States of

matter and

intermolecular forces |

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Pure Substances and
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Elements \u0026

Compounds.

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atmosphere, ozone
formation,
chlorofluorocarbons,
CFC ' s 1.2 Chemistry
and Matter Chemistry
Central Science

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Chapter 1 Section 2
MATTER and Its
Properties Matter –

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Chemistry

Anything that has MASS and VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest unit of an element that still maintains the properties that element.

ELEMENT – a pure substance made up of only one type of atom.

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that best completes the statement or answers the question. 1. Atomic radii cannot be measured ...

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Solutions Manual

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Section 3.1 Properties of
Matter pages 70–75

Problem-Solving Lab 1.
Explain why the flow of a
compressed gas must be
controlled for practical
and safe use.

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Changes Matter—Proper
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The Periodic Table and
Periodic Law Section 6.1

Development of the
Modern Periodic Table

1. octaves 2. eight 3. nine

4. accepted 5. Dmitri

Mendeleev 6. atomic

mass 7. elements 8.

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atomic number 9. Henry

Moseley 10. protons 11.

periodic law 12.

properties 13. 14 ...

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Atoms Reviewing

Vocabulary Match the

definition in Column A

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with the term in Column

B. Column A Column B

1. The set of frequencies
of the electromagnetic

waves emitted by the

atoms of an element 2.

The minimum amount of
energy that can be lost or
gained by ...

Chapter Assessment

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intensive property.

extensive properties

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the amount of matter in a

sample. a property that

depends on the type of

matter in a sample not....

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mass, volume, length,
weight.

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moles of known

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H O H O number of

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moles of unknown
substance O CO H O C
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 C H O CO use mole
ratio $5 \text{ mol O} : 2 \text{ mol C} : 3$
 $\text{H} : 8 \text{ mol CO} : 2 \text{ mol C}$
 $3 \text{ H} : 8 \text{ mol H}_2\text{O} : 1 \text{ mol}$
 $\text{C} : 3 \text{ H} : 8 \text{ mol C} : 3 \text{ H} : 8 \text{ mol}$
 $\text{O} : 2$

MATH SKILLS
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A “ chemical change ”

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is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds Quite different properties than their component

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